

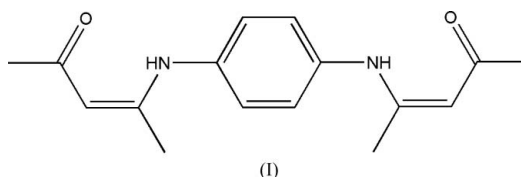
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Key indicators

Single-crystal X-ray study
 $T = 295$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.037
 wR factor = 0.101
Data-to-parameter ratio = 16.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.4,4'-(*p*-Phenylenediimino)dipent-3-en-2-oneThe molecule of the title Schiff base compound, $\text{C}_{16}\text{H}_{20}\text{N}_2\text{O}_2$, lies on a crystallographic twofold rotation axis. The crystal structure is stabilized by weaker intermolecular $\text{C}-\text{H}\cdots\text{O}$ and stronger intramolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds.Received 9 March 2007
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Comment

The title compound, (I), has been reported to be a good polydentate ligand which can be a building block for larger molecules and supramolecular assemblies and coordinate with various metals with different properties (Shauib *et al.*, 2006). It can be obtained *via* two different methods in high yield (Shauib *et al.*, 2006). We obtained (I) in over 90% yield from the reaction of *p*-phenylenediamine with two equivalents of 2,4-pentanedione *via* a method slightly different from those in the literature (Shauib *et al.*, 2006) (see *Experimental*).A crystallographic twofold rotation axis passes through the mid-points of the $\text{C}7-\text{C}7\text{A}$ and $\text{C}8-\text{C}8\text{A}$ bonds (Fig. 1). The title compound is generally believed to exist in three main forms: dione–diimine, dione–dienamine and dienol–diimine (Shauib *et al.*, 2006). The crystal structure indicates that (I) exists in the dione–dienamine form in the solid state, according to the bond lengths $\text{C}2-\text{O}1$ [1.2528 (15) Å], $\text{C}2-\text{C}3$ [1.4285 (17) Å], $\text{C}3-\text{C}4$ [1.3813 (17) Å] and $\text{C}4-\text{N}1$ [1.3548 (15) Å]. Atoms $\text{C}2$, $\text{C}3$, $\text{C}4$, $\text{O}1$ and $\text{N}1$ are nearly coplanar. The $\text{N}1-\text{C}4-\text{C}3-\text{C}2$ and $\text{O}1-\text{C}2-\text{C}3-\text{C}4$ torsion angles are 0.45 (19) and -2.9 (2)°, respectively. The dihedral angle between the benzene plane $\text{C}6/\text{C}6\text{a}/\text{C}7/\text{C}7\text{a}/\text{C}8/\text{C}8\text{a}$ and the plane formed by atoms $\text{C}2/\text{C}3/\text{C}4/\text{O}1/\text{N}1$ is 45.9° .The crystal structure (Fig. 2) is stabilized by weaker intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds (the upper limit of the $\text{C}\cdots\text{O}$ distance in $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds is 4.0 Å; Desiraju, 1996; Steed & Atwood, 2000), forming an S-shaped sheet, and stronger intramolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 1).

Experimental

A mixture of 2,4-pentanedione (5.00 ml, 48.9 mmol), *p*-phenylenediamine (2.64 g, 24.5 mmol) and catalytic concentrated HCl (1% mol ratio) was refluxed in absolute ethanol (20 ml). The crude product precipitated from the solution and was recrystallized from ethanol to give yellow crystals (yield 6.07 g, 91%).

Crystal data

 $C_{16}H_{20}N_2O_2$ $M_r = 272.34$ Orthorhombic, *Pbcn* $a = 10.675 (2) \text{ \AA}$ $b = 8.8373 (18) \text{ \AA}$ $c = 14.995 (3) \text{ \AA}$ $V = 1414.5 (5) \text{ \AA}^3$ $Z = 4$ Mo $K\alpha$ radiation $\mu = 0.09 \text{ mm}^{-1}$ $T = 295 (2) \text{ K}$ $0.1 \times 0.1 \times 0.1 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer

1610 independent reflections

Absorption correction: none

1343 reflections with $I > 2\sigma(I)$

12933 measured reflections

 $R_{\text{int}} = 0.043$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.037$

H atoms treated by a mixture of independent and constrained refinement

 $wR(F^2) = 0.101$ $S = 1.06$

1610 reflections

97 parameters

 $\Delta\rho_{\text{max}} = 0.25 \text{ e \AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.22 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C7-H7\cdots O1^i$	0.93	3.08	3.9803 (16)	163
$N1-H1\cdots O1$	0.920 (18)	1.913 (17)	2.6776 (14)	139.3 (15)

Symmetry code: (i) $-x + 1, -y, -z + 1$.

C-bound H atoms were positioned geometrically, with $C-H = 0.93-0.96 \text{ \AA}$, and allowed to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5U_{\text{eq}}(\text{C})$. Atom H1 was located in a difference Fourier map and refined freely with an isotropic displacement parameter.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1998); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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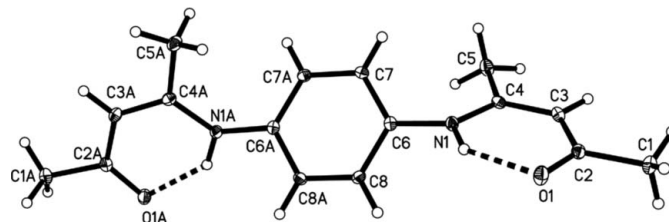


Figure 1

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented by spheres of arbitrary radius. Intramolecular hydrogen bonds are indicated by dashed lines. [Symmetry code: (A) $1 - x, y, \frac{3}{2} - z$.]

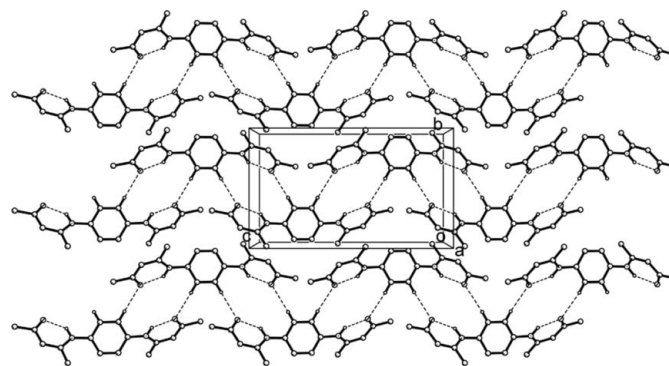


Figure 2

The packing of (I). Hydrogen bonds are indicated by dashed lines.

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